Thiophen Derivatives of Biological Interest. Part IX.* Odour and Chemical Structure in the Thiophen Series.

By MICHEL SY, NG. PH. BUU-HoI, and NG. D. XUONG.

[Reprint Order No. 5720.]

The relation between odour and chemical constitution in thiophen compounds has been investigated. 2- and 3-tert.-Butyl-4:5:6:7-tetra-hydro-4-oxothionaphthen were synthesised from 2- and 3-tert.-butylthiophen by the succinic anhydride method; their vapours have an odour reminiscent of 7-tert.-butyl-1-tetralone; 5-isopropyl-2-thenaldehyde, 2-2'-thienylideneheptanal, and a series of esters of 2-thenoic acid have odours similar to those of the benzene isologues. Other reactions of 3-tert.-butylthiophen and 2:5-di-tert.-butylthiophen were investigated.

7-tert.-BUTYL-1-TETRALONE has an odour reminiscent of smouldering sandalwood (Buu-Hoī and Cagniant, Compt. rend., 1942, 214, 115; Cagniant and Buu-Hoī, Bull. Soc. chim., 1942, 9, 111, 841). It was of interest to examine the odour of its isologues. 2-tert.-Butyl-4:5:6:7-tetrahydro-4-oxothionaphthen (I), prepared from 2-tert.-butylthiophen by the succinic anhydride method (cf. Buu-Hoī, Hoán, and Khôi, J. Org. Chem., 1950, 15, 957), did in fact, in the vapour state, have the characteristic burning-sandalwood smell. The 3-tert.-butyl isomer, prepared from 3-tert.-butylthiophen, had a less marked and less pleasant odour. A chemical difference was that the former readily underwent a Pfitzinger reaction with isatin to give 5'-tert.-butyl-3:4-dihydrothiopheno(3':2'-1:2)acridine-5-carboxylic acid (III), whereas the latter gave a negligible yield, probably on account of steric hindrance by the tert.-butyl group.

The successful synthesis of the thionaphthen (II) showed that succinoplation of 3-tert.-butylthiophen, which yielded only one reaction product, occurred at position 5, in contrast with the acylation of 3-methylthiophen which takes place mostly at position 2 (Hartough and Kosak, J. Amer. Chem. Soc., 1947, 69, 3093), a discrepancy probably due to the steric effect of the tert.-butyl radical. Formylation of 3-tert.-butylthiophen readily gave a single aldehyde, which was therefore taken to be 4-tert.-butyl-2-thenaldehyde; likewise, acetylation gave a single ketone, considered on the same grounds to be 2-acetyl-4-tert.-butyl-thiophen. This ketone underwent a Pfitzinger reaction with isatin to give 2-(4-tert.-butyl-2-thienyl)cinchoninic acid, and oxidation with sodium hypobromite to give 4-tert.-butyl-2-thenoic acid.

Formylation of 2:5-di-tert.-butylthiophen gave a good yield of 2:5-di-tert.-butyl-3-thenaldehyde; condensation with benzyl cyanide gave the acrylonitrile (IV; R = H). 2:5-Di-tert.-butyl-3-methylthiophen, prepared from the aldehyde by Wolff-Kishner reduction, also gave the aldehyde, and thence the acrylonitrile (IV; R = Me) and 2:5-di-tert.-butyl-3:4-dimethylthiophen. 3-Acetyl-2:5-di-tert.-butylthiophen, unlike the aldehydes, gave no ketonic derivatives or cinchoninic acid, nor could it be oxidised by sodium hypobromite.

5-isoPropyl-2-thenaldehyde was prepared; its odour was like that of *p-iso*propyl-benzaldehyde. Both 2-2'-thienylideneheptanol, prepared by condensing 2-thenaldehyde with heptanal, and α-n-pentylcinnamaldehyde (Rutowski and Korolew, *J. prakt. Chem.*,

1928, 119, 273) had a jasmin-like scent; similarly, a series of esters prepared from 2-thenoic acid had odours resembling those of their benzene isologues.

There is thus an analogy between the benzene and thiophen derivatives in respect of odour; but no attempt was made in the present work at a precise assessment on a physiological basis.

EXPERIMENTAL

 β -(5-tert.-Butyl-2-thenoyl) propionic Acid.—To an ice-cooled, stirred solution of 2-tert.-butylthiophen (22 g.) and succinic anhydride (20 g.) in nitrobenzene (120 c.c.), finely powdered aluminium chloride (50 g.) was added portionwise. The mixture was kept for 16 hr. at room temperature, then decomposed with ice, and the nitrobenzene steam-distilled. The keto-acid was taken up in aqueous sodium carbonate and recrystallised from benzene, giving colourless prisms (31 g.), m. p. 114° (Found: C, 59·7; H, 6·8. $C_{12}H_{16}O_3S$ requires C, 60·0; H, 6·7%).

 γ -(5-tert.-Butyl-2-thienyl)butyric Acid.—The foregoing acid (28 g.) was heated with 85% hydrazine hydrate (30 g.) and potassium hydroxide (25 g.) in diethylene glycol (100 c.c.) to $190-200^\circ$ with removal of water, and the mixture then refluxed for 2 hr., diluted with water, and acidified with dilute hydrochloric acid. The reduction product, taken up in benzene, was purified by vacuum-distillation, and formed colourless leaflets (20 g.), m. p. 47°, b. p. $214^\circ/24$ mm., from ligroin (Found: C, 63·5; H, 8·2. $C_{12}H_{18}O_2S$ requires C, 63·7; H, 8·0%).

2-tert.-Butyl-4:5:6:7-tetrahydro-4-oxothionaphthen (I).—The butyric acid (10 g.) was converted into the acid chloride by thionyl chloride (20 g.) and a few drops of pyridine, treated in cold carbon disulphide (50 c.c.) with stannic chloride (25 g., dissolved in 20 c.c. of carbon disulphide), refluxed for 2 hr., and worked up in the usual way. The ketone (4 g.) formed a pale yellow oil, b. p. $170-171^{\circ}/13$ mm., n_{2}^{5} 1.5699 (Found: C, 68.9; H, 7.8. $C_{12}H_{16}OS$ requires C, 69.2; H, 7.7%). The semicarbazone formed needles, m. p. 249° , from ethanol (Found: N, 15.6. $C_{13}H_{19}ON_{3}S$ requires N, 15.8%).

5'-tert.-Butyl-3: 4-dihydrothiopheno(3': 2'-1: 2)acridine-5-carboxylic Acid (III).—A solution of the foregoing ketone (2·1 g.), isatin (1·5 g.), and potassium hydroxide (1·7 g.) in ethanol (15 c.c.) was gently refluxed for 12 hr. After dilution with water and ether-extraction of neutral impurities, acidification with acetic acid gave a precipitate (2 g.) which crystallised as pale yellow prisms, m. p. 231°, from ethanol (Found: C, 71·0; H, 5·6. $C_{20}H_{19}O_2NS$ requires C, 71·2; H, 5·6%).

4-tert.-Butyl-2-thenaldehyde.—A mixture of 3-tert.-butylthiophen (17·5 g.), N-methylformanilide (22·5 g.), and phosphorus oxychloride (24 g.) was refluxed for 2 hr., then poured into aqueous sodium acetate, and the mixture was steam-distilled. The aldehyde was taken up in ether, washed with dilute hydrochloric acid, then with water, dried (Na₂SO₄), and purified by distillation, to form a pale yellow oil (18 g.), b. p. 249—250°, n_D² 1·5320 (Found: C, 64·2; H, 7·2. C₉H₁₂OS requires C, 64·3; H, 7·1%). It gave a semicarbazone, leaflets, m. p. 221°, from ethanol (Found: N, 18·4. C₁₀H₁₅ON₃S requires N, 18·7%), thiosemicarbazone, colourless prisms, m. p. 186°, from ethanol (Found: C, 49·5; H, 6·2. C₁₀H₁₅N₃S₂ requires C, 49·8; H, 6·2%), 4-oxo-Δ²-thiazolin-2-ylhydrazone, colourless prisms, m. p. 315°, from ethanol (Found: C, 51·0; H, 5·2. C₁₂H₁₅ON₃S₂ requires C, 51·2; H, 5·3%), 2:4-dinitrophenylhydrazone, red needles, m. p. 221°, from acetic acid (Found: N, 16·2. C₁₅H₁₆O₄N₄S requires N, 16·1%), and oxime, colourless needles, m. p. 79°, from methanol (Found: C, 58·8; H, 7·4. C₉H₁₃ONS requires C, 59·0; H, 7·1%).

4-tert.-Butyl-2-methylthiophen.—Obtained by reduction of the foregoing aldehyde (8·5 g.) by means of hydrazine hydrate (14 g.) and potassium hydroxide (14 g.) in diethylene glycol (60 c.c.), this compound formed a pale yellow oil (7 g.), b. p. 195°, n_D^{24} 1·5025 (Found: C, 70·0; H, 9·2. $C_9H_{14}S$ reuires C, 70·1; H, 9·1%).

2-Acetyl-4-tert.-butylthiophen.—To an ice-cooled solution of 3-tert.-butylthiophen (14 g.) and acetyl chloride (8 g.) in carbon disulphide (200 c.c.), stannic chloride (30 g.) was added portionwise. The mixture was kept at room temperature for 1 hr., poured on ice, then worked up in the usual way. The ketone (15 g.) was a pale yellow oil, b. p. 259°, n_{23}^{20} 1·5322 (Found C, 65·8; H, 7·7. C₁₀H₁₄OS requires C, 65·9; H, 7·7%). It gave a 2:4-dinitrophenylhydrazone, red needles, m. p. 226°, from acetic acid (Found: N, 15·8. C₁₆H₁₈O₄N₄S requires N, 15·5%), oxime, colourless needles, m. p. 113°, from ethanol (Found: C, 60·6; H, 7·5. C₁₀H₁₅ONS requires C, 60·9; H, 7·6%), semicarbazone, m. p. 249°, thiosemicarbazone, m. p. 192°, and 4-oxo-Δ²-thiazolin-2-ylhydrazone, m. p. 305°. 2-(4-tert.-Butyl-2-thienyl)cinchoninic acid, prepared in 90% yield by means of isatin, formed pale yellow prisms, m. p. 172°, from ethanol (Found: C, 69·1; H, 5·3. C₁₈H₁₇O₂NS requires C, 69·5; H, 5·5%).

4-tert.-Butyl-2-thenoic Acid.—2-Acetyl-4-tert.-butylthiophen (19 g.) was oxidised with aqueous sodium hypobromite (from 16 g. of sodium hydroxide and 8.2 c.c. of bromine); the acid obtained (60% yield) formed prisms, m. p. 77°, from water (Found: C, 58.5; H, 6.6. $C_0H_{12}O_2S$ requires C, 58.7; H, 6.5%).

 β -(4-tert.-Butyl-2-thenoyl) propionic Acid.—Prepared from 3-tert.-butylthiophen as for the 2-isomer and in similar yield, this acid formed colourless prisms, m. p. 119°, from benzene (Found: C, 59·7; H, 6·5%). γ -(3-tert.-Butyl-5-thienyl) butyric acid formed a pale yellow, viscous oil, b. p. 200—201°/13 mm. (Found: C, 63·5; H, 8·3%).

3-tert.-Butyl-4:5:6:7-tetrahydro-4-oxothiophen.—This ketone formed a pale yellow oil, b. p. $179^{\circ}/13$ mm., $n_{\rm p}^{26}$ 1:5669, whose vapours had a less pleasant odour than the isomeric ketone (Found: C, 69.0; H, 8.0%); it gave no cinchoninic acid with isatin. The semicarbazone formed needles, m. p. 238° (Found: C, 58.6; H, 7.3%).

2:5-Di-tert.-butyl-3-thenaldehyde.—A mixture of 2:5-di-tert.-butylthiophen (25 g.), phosphorus oxychloride (25 g.), and N-methylformanilide (24 g.) was heated for 3 hr. on the waterbath and worked up in the usual way; the aldehyde formed a pale yellow oil, b. p. $137^{\circ}/13$ mm., $n_2^{\rm D}$ 1·5147 (Found: C, 69·5; H, 9·0. $C_{13}H_{20}{\rm OS}$ requires C, 69·6; H, 8·9%), and gave a semicarbazone, prisms, m. p. 231° (Found: N, 14·6. $C_{14}H_{23}{\rm ON}_3{\rm S}$ requires N, 14·9%), oxime, prisms, m. p. 106°, from ethanol (Found: C, 65·0; H, 9·0. $C_{13}H_{21}{\rm ONS}$ requires C, 65·3; H, 8·8%), and 2:4-dinitrophenylhydrazone, red needles, m. p. 219°, from acetic acid (Found: N, 14·2. $C_{19}H_{24}O_4{\rm N}_4{\rm S}$ requires N, 13·9%).

1-Phenyl-2-(2:5-di-tert.-butyl-3-thienyl)acrylonitrile (IV; R=H).—This compound crystallised as pale yellow needles, m. p. 136°, from ethanol, giving a deep orange solution with sulphuric acid (Found: C, 78-2; H, 7-6. $C_{21}H_{25}NS$ requires C, 78-0; H, 7-7%).

2:5-Di-tert.-butyl-3-methylthiophen.—Prepared from 2:5-di-tert.-butyl-3-thenaldehyde (10 g.), hydrazine hydrate (15 g.), and potassium hydroxide (15 g.) in diethylene glycol (50 c.c.), this compound formed a pale yellow oil (8 g.), b. p. $119^{\circ}/13$ mm., $n_{\rm p}^{23}$ 1·4952 (Found: C, 74·2; H, 10·5. $C_{13}H_{22}S$ requires C, 74·3; H, $10\cdot5\%$).

2:5-Di-tert.-butyl-4-methyl-3-thenaldehyde.—Prepared from the foregoing compound (8 g.), phosphorus oxychloride (8 g.), and N-methylformanilide (7 g.), the aldehyde (7 g.) was a pale yellow, viscous oil, b. p. $155^{\circ}/13$ mm., n_D^{21} 1·5091 (Found: C, $70\cdot4$; H, $9\cdot5$. $C_{14}H_{22}OS$ requires C, $70\cdot6$; H, $9\cdot2\%$); it gave a semicarbazone, needles, m. p. 225° , from ethanol (Found: N, $13\cdot9$. $C_{15}H_{25}ON_3S$ requires N, $14\cdot2\%$), oxime, prisms, m. p. 95° , from ethanol (Found: C, $66\cdot3$; H, $9\cdot3$. $C_{14}H_{23}ONS$ requires C, $66\cdot4$; H, $9\cdot1\%$), and orange-red 2:4-dinitrophenyl-hydrazone, m. p. 221° .

2-(2:5-Di-tert.-bulyl-4-methyl-3-thienyl)-1-phenylacrylonitrile (IV; R = Me).—This compound crystallised as yellow needles, m. p. 144° from ethanol, giving an orange-red solution with sulphuric acid (Found: C, 78·0; H, 8·3. $C_{22}H_{27}NS$ requires C, 78·3; H, 8·0%).

2:5-Di-tert.-butyl-3:4-dimethylthiophen.—Obtained in only 60% yield, this compound was a pale yellow oil, b. p. 139°/13 mm., $n_{\rm D}^{19}$ 1·4832 (Found: C, 74·9; H, 10·9. $C_{14}H_{24}S$ requires C, 75·0; H, 10·7%).

5-isoPropyl-2-thenaldehyde.—Prepared in the usual way from 2-isopropylthiophen (Schleicher, Ber., 1886, 19, 672) (b. p. 154°, n_D^{20} 1·5630), the aldehyde (78% yield) was a pale yellow oil, b. p. 239°, n_D^{19} 1·5604, with a strong cumin odour resembling that of p-isopropylbenzaldehyde (Found: C, 62·3; H, 6·5. C_8H_{10} OS requires C, 62·3; H, 6·5%). The semicarbazone formed needles, m. p. 219°, from ethanol (Found: C, 51·0; H, 6·3. $C_9H_{13}ON_3S$ requires C, 51·2; H, 6·2%), the thiosemicarbazone yellowish prisms, m. p. 203°, from ethanol (Found: C, 47·3; H, 5·5. $C_9H_{13}N_3S_2$ requires C, 47·6; H, 5·7%); and the 4-oxo- Δ^2 -thiazolin-2-ylhydrazone colourless needles, m. p. 257° (decomp.) (Found: C, 49·1; H, 5·0. $C_{11}H_{13}ON_3S_2$ requires C, 49·4; H, 4·9%). 1-Phenyl-2-(5-isopropyl-2-thienyl)acrylonitrile, obtained by use of benzyl cyanide, formed yellowish needles, m. p. 72°, from ethanol, giving an orange colour with sulphuric acid (Found: C, 75·8; H, 6·2. $C_{18}H_{15}NS$ requires C, 75·9; H, 5·9%).

2-2'-Thienylideneheptanal.—To a solution of 2-thenaldehyde (11·2 g.) and heptanal (11·4 g.) in ethanol (100 c.c.) and water (500 c.c.), 10% aqueous sodium hydroxide (25 c.c.) was added, and the mixture was kept at room temperature for 2 days, with frequent shaking. After acidification with acetic acid, the product was taken up in ether and vacuum-fractionated, giving a colourless oil (15 g.), b. p. 192—194°/35 mm., n_2^{25} 1·5686 (Found: C, 69·0; H, 7·8. C₁₂H₁₆OS requires C, 69·2; H, 7·7%). The semicarbazone formed colourless prisms, m. p. 132—133°, from ethanol (Found: C, 58·6; H, 7·4. C₁₃H₁₉ON₃S requires C, 58·9; H, 7·2%).

Esters of 2-Thenoic and 5-tert.-Butyl-2-thenoic Acid.—These esters were prepared by the

reaction of 2-then oyl chloride or 5-tert.-butyl-2-thenoyl chloride on the appropriate alcohol in ether or benzene; they are listed in the Table.

Alkyl	B n lmm	# (tomm)	Formula				Reqd. (%)	
Aikyi		` - /	Formula	С	Н	С	H	
Esters of 2-thenoic acid.								
Pr ⁿ	$124-125^{\circ}/24$	$1.5137 (28^{\circ})$	$C_8H_{10}O_2S$	56.3	6.1	56.5	5.9	
Pr ^t	110°/22	1.5128 (26)	$C_8H_{10}O_2S$	56.2	6.0	56.5	5.9	
Bu ⁿ	137138°/24	1.5112(27)	C ₉ H ₁₂ O ₃ S	58.8	6.6	58.7	6.5	
Bu ^t	129°/22	1.5088(27)	$C_{\bullet}H_{1\bullet}O_{\bullet}S$	58.4	6.7	58.7	6.5	
n-Pentyl	152-153°/27	1.5079(27)	$C_{10}H_{14}O_{\bullet}S$	60.5	7.3	60.6	7.1	
isoPentyl	142143°/22	1·5049 (28)	$C_{10}^{10}H_{14}^{14}O_{2}S$	60.8	7.0	60.6	$7 \cdot 1$	
n-Decyl	207208°/20	1·4943 (26)	$C_{15}H_{24}O_{2}S$	67.0	8.8	67.2	9.0	
Tetrahydrogeranyl	198°/22	1.4986(27)	$C_{15}H_{24}O_{2}S$	67.3	$9 \cdot 2$	67.2	9.0	
n-Dodecyl	$224-225^{\circ}/20$	1.4901(28)	$C_{17}H_{28}O_{2}S$	68.6	9.4	68.9	9.5	
n-Tetradecyl	$245-246^{\circ}/20$	1.4891(27)	$C_{19}H_{32}O_{2}S$	70.3	10.2	70.4	9.9	
n-Hexadecyl	266268°/22	1·4882 (26)	$C_{21}H_{36}O_{2}S$	71.3	10.2	71.6	10.2	
Citronellyl	198199°/20	1·4981 (26)	C ₁₅ H ₂₂ O ₂ S	67.5	8.5	67.7	8.3	
Benzyl	195196°/24	1.5786(25)	$C_{12}H_{10}O_{2}S$	66.0	4.7	66.1	4.6	
2-Phenylethyl	199201°/18	1.5731 (26)	C ₁₃ H ₁₂ O ₂ S	67.0	5.5	67.2	5.2	
3-Phenylpropyl	222°/24	1.5649(26)	C14H14O.S	68.2	5.8	68.3	5.7	
	/	()	014-114040		0 0	•••	•	
Esters of 5-tertbutyl-2-thenoic acid.								
Et	150151°/20	1.5138 (26)	$C_{11}H_{16}O_{2}S$	62.0	7.3	$62 \cdot 3$	7.5	
isoPentyl	174—175°/20	1.5025(25)	C ₁₄ H ₂ ,O ₂ S	66.0	8.8	66.1	8.7	
Benzyl	226—227°/29	1.5608(27)	$C_{16}H_{18}O_{2}S$	70.3	6.6	70.1	6.6	
2-Phenylethyl	238239°/31	1.5493 (27)	C ₁₇ H ₂₀ O ₂ S	70.5	7.2	70.8	6.9	
True Barrers Incommune Harrison on Danie				D 1	C. 141	10.7	10541	

THE RADIUM INSTITUTE, UNIVERSITY OF PARIS.

[Received, September 13th, 1954.]